Phase Determination of Black TiO₂ Nanoparticles

Meng-kun.Tian,¹ Masoud. Mahjouri-Samani,² Gyula. Eres,² Ritesh.Sachan,³ Matthew F. Chisholm³, Kai Wang,² Alexander. A. Puretzky,² Christopher. M. Rouleau, ² Mina. Yoon, ²David. B. Geohegan,² and Gerd Duscher^{1,3}

1.Dept. of Materials Science and Engineering, University of Tennessee, Knoxville, TN, USA

2. Center for Nanophase Materials Sciences, Oak Ridge National Laboratory, Oak Ridge, TN, USA

3. Material Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge, TN, USA

The increased photocatalytic activity of black TiO_2 nanoparticle [1,2] is attributed to the bandgap narrowing in the surface layer produced by crystallization of amorphous material in a reducing atmosphere. These nanoparticles were found to have unique crystalline core-amorphous shell structures of TiO_2 stoichiometry. Here we report a form of black TiO_2 nanoparticles with different core-shell structure characterized by highly spatially resolved characterization methods. These particles were transformed from amorphous materials by post annealing. Spectra of monochromated electron energy-loss spectroscopy (EELS) show significant deviations of reference rutile $Ti-L_{2,3}$ and O-K energy-loss near edge structure (ELNES) in the near surface area (2.5nm-1nm away from vacuum) of a rutile nanoparticle, while no clear deviations are observed in the interior (7nm-4nm away from vacuum). Through a linear combination and linear least-square fits of these spectra, we find that deviation of ELNES originates from high concentration of Ti_2O_3 . This result indicates a stoichiometric rutile core- Ti_2O_3 shell structure.

Fig 1a shows the HRTEM image of black TiO₂ (rutile with diameter>15nm) nanoparticle. EELS spectra were taken from 7nm to 1nm away from surface along the direction highlighted by the arrow. As shown in Fig 1b, spectrum taken from 7nm to 4nm looks similar with each other, showing rutile characteristic. However, a significant changes of the EELS spectra taken at 2.5 nm and 1nm is evident: the two L_3 -eg peaks have roughly the same maximum intensity (2.5nm away from vacuum), but ELNES from spectra taken at 1nm away from vacuum further changed into smeared out anatase signature. These smeared out feature have been attribute to the existence of Ti^{3+} component [3, 4]. In order to explore these structure changes quantitatively, we use the linear combinations of pure rutile and pure Ti₂O₃ reference EELS spectra to fit the experimental data, as shown in Fig 1c. We use the linear least-square fitting to find the best fit between experimental and these reconstructed spectra yielding the TiO₂ concentrations as coefficients. The fit errors are very close to the noise level (max<13%). From the fit, Ti₂O₃ component in this nanoparticles are responsible for the deviations of rutile ELNES: 33% Ti₂O₃ at 2.5nm, 60% Ti₂O₃ at 1nm. Even though deviations of Ti-L_{2.3} ELNES are not easily observed in the range of 7nm-4nm, the least-square fits show a low Ti₂O₃ concentration in the particle inner (11%-18%). This is consistent with a core-shell structure.

The ratio of the maximum intensity of the first two peaks (a and b for Ti_2O_3 , and a' and b' for TiO_2) uniquely identifies Ti_2O_3 and TiO_2 . Specifically, for Ti_2O_3 a/b<1, while for TiO_2 a/b>1, as

shown in Fig 2a. No obvious changes were observed in O-K ELNES at each location except near surface area where the high Ti_2O_3/TiO_2 ratio gives rise to a slight change in the ratio of the maximum intensities of the first two peaks in the O-K edge at 2.5nm and a significant change 1nm away from the vacuum, as shown in Fig 2b and f. Ti_2O_3 coefficients from fitting the O-K ELNES are consistent with those of $Ti-L_{2,3}$ ELNES within a 2% error at each location except 1nm where signal-to-noise ratio of O-K edges signals is high. Thus, the fit of ELNES is highly reliable, showing a Ti_2O_3 shell and stoichiometric rutile core of a form of black TiO_2 nanoparticles not reported previously.

References

- [1]. X. B Chen et al. Science 331 (2011), p. 746-750.
- [2] A. Naldoni et al. J Am Chem Soc 134 (2012), p. 7600-7603.
- [3] A. S.Sefat, et al. J Solid State Chem 178 (2005), p. 1008-1016.
- [4] A.Ohtomo, A. et al. Nature 419 (2002), p. 378-380.

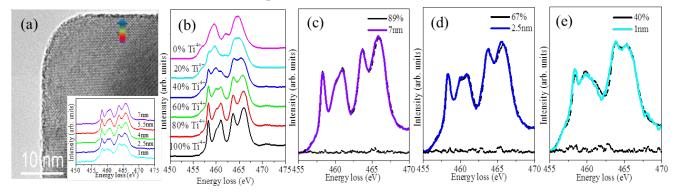


Figure 1(a) HRTEM image of rutile particle viewed in [001] zone axis. EELS spectra (inset) taken from particle inner to surface along the direction highlighted by the arrow. (b) Linear combination of $Ti-L_{2,3}$ ELNES from pure rutile and Ti_2O_3 at different ratios. (c)-(e) Linear least-square fit of EELS spectra taken from 7nm (represent for particle inner), 2.5nm and 1nm. The difference between the experimental and reconstructed EELS spectra are given at the bottom of each image.

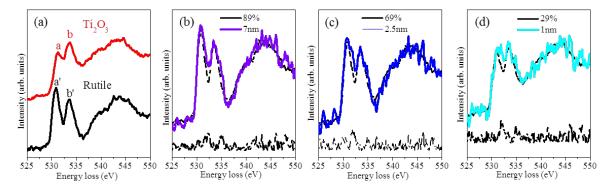


Figure 2 (a) O-K ELNES from pure rutile and Ti_2O_3 . (c)-(e) Linear least-square fit of EELS spectra taken from 7nm, 2.5nm and 1nm. Difference between the experimental and reconstructed EELS spectra are given at the bottom of each image.